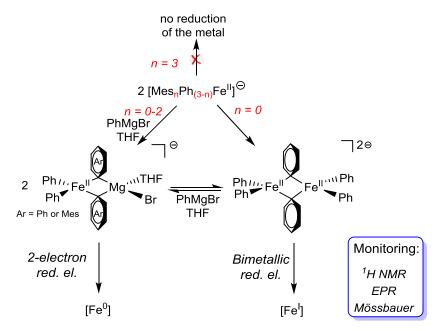
## Formation of low-valent Fe<sup>0</sup> and Fe<sup>1</sup> species in Fe-catalyzed cross-coupling chemistry: key role of ate-Fe<sup>II</sup> intermediates

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Ate-iron(II) species such as  $[Ar_3Fe^{II}]^-$  (Ar = aryl) are key intermediates in Fe-catalyzed cross-coupling reactions between aryl Grignard reagents (ArMgX) and organic electrophiles.<sup>[1]</sup> They can be active species in the catalytic cycle,<sup>[2]</sup> or lead to Fe<sup>0</sup> and Fe<sup>I</sup> oxidation states. These low oxidation states were shown to be catalytically active in some cases, but they mostly lead to unwished organic byproducts.<sup>[3,4]</sup>

This works relates a study of the evolution of [Ar<sub>3</sub>Fe<sup>II</sup>]<sup>-</sup> complexes towards Fe<sup>0</sup> and Fe<sup>I</sup> oxidation states, through <sup>1</sup>H NMR, EPR and <sup>57</sup>Fe-Mössbauer spectroscopies, as well as DFT calculations, so as to discuss the role of both steric parameters and spin states in the reduction processes. Such mechanistic insights give a better understanding of iron-catalyzed C-C bond formation reactions, and can be exploited in the design of new ligands in order to selectively obtain a sole iron oxidation state in a catalytic process.



**Figure 1:** Pathways connecting ate-Fe<sup>II</sup> (Mes = 2,4,6-trimethylphenyl) species with Fe<sup>0</sup> and Fe<sup>I</sup> oxidation states

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